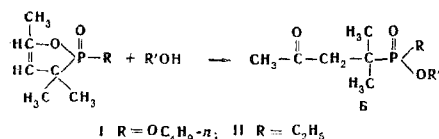


ketophosphonic and ketophosphinic acids. The reaction takes place in accordance with the following equation.



The products and their constants are given in the table. The initial cyclic ester I was obtained by the reaction of butanol with 2-chloro-3,3,5-trimethyl-2-oxo-1,2-oxaphosphol-4-ene [4] in the presence of triethylamine with a yield of 38%. Bp 127°–129° C (9 mm);  $d_4^{20}$  1.0390;  $n_D^{20}$  1.4535. Found, %: P 14.27; MR<sub>D</sub> 56.81. Calculated for C<sub>10</sub>H<sub>19</sub>O<sub>3</sub>P, %: P 14.19; MR<sub>D</sub> 56.58. The cyclic compound II was obtained from ethyldichlorophosphine, mesityl oxide, and acetic anhydride in a manner similar to that described by Bergesen [2] with a yield of 48%. Bp 69°–71° C (0.1 mm);  $d_4^{20}$  1.0565;  $n_D^{20}$  1.4757. Found, %: C 55.54; H 8.55; P 17.45; MR<sub>D</sub> 46.34. Calculated for

C<sub>8</sub>H<sub>15</sub>O<sub>2</sub>P, %: C 55.16; H 8.68; P 17.78; MR<sub>D</sub> 46.22. The structure of the products was confirmed by their spectra and chemical reactions.

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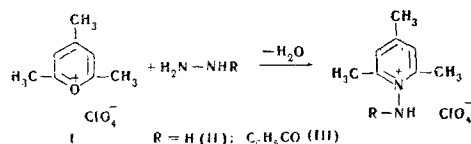
## CONVERSION OF PYRYLIUM SALTS INTO 1-AMINOPYRIDINIUM AND PYRAZOLE DERIVATIVES

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It is known that the reaction of pyrylium salts with arylhydrazines gives 1-arylamino-pyridinium salts, which are converted by the action of bases into pyridinium betaines [1]. In order to obtain 1-aminopyridinium compounds, we have studied the reaction of pyrylium salts with hydrazine. We have found that the reaction of 2,4,6-trimethylpyrylium perchlorate (I) with an equimolecular amount of hydrazine hydrate takes place in ethanolic solution with cooling to form 1-aminocollidinium perchlorate (II) with a yield of 24–25%.

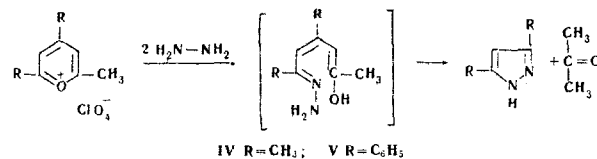


Compound II forms colorless needles soluble in water and ethanol with mp 185° C (from ethanol). Found, %: C 40.83; H 5.48; Cl 15.32. Calculated for C<sub>8</sub>H<sub>13</sub>ClN<sub>2</sub>O<sub>4</sub>, %: C 40.60; H 5.53; Cl 14.98. 1-Benzoylamino-collidinium perchlorate (III) was obtained by boiling benzoylhydrazine with I in glacial acetic acid with a yield of 73%. Colorless crystals readily soluble in ethanol, sparingly soluble in water, mp 147° C (from water). Found, %: C 52.38; H 5.08; Cl 10.23. Calculated for C<sub>15</sub>H<sub>17</sub>ClN<sub>2</sub>O<sub>5</sub>, %: C 52.87; H 5.02; Cl 10.40.

The reaction of I with a twofold amount of hydrazine hydrate yielded a crystalline product IV containing no halogen to which the structure of 3,5-dimethylpyrazole has been assigned on the basis of its analysis and IR spectrum. Colorless plates soluble in benzene, ethyl acetate, and water, with mp 106°–107° C (from ethyl acetate). Found, %: C 62.57; H 8.44. Calculated for C<sub>5</sub>H<sub>8</sub>N<sub>2</sub>, %: C 62.46; H 8.38. The IR spectrum of IV and that of an authentic sample of 3,5-dimethyl-

pyrazole were completely identical, and the mixture gave no depression of the melting point.

Obviously, in this case the intermediate 1,5-diketone hydrazone undergoes cleavage with subsequent cyclization to form a pyrazole compound.



6-Methyl-2,4-diphenylpyrylium perchlorate and hydrazine hydrate (1:2) yielded 3,5-diphenylpyrazole (V). Crystalline product, soluble in ethanol and benzene. Mp 200° C (from benzene). Found, %: C 82.39; H 5.48. Calculated for C<sub>15</sub>H<sub>12</sub>N<sub>2</sub>, %: C 81.81; H 5.49. The structure of the product was shown in a manner similar to that of IV.

The investigation of the reactions of pyrylium salts with hydrazine is continuing.

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